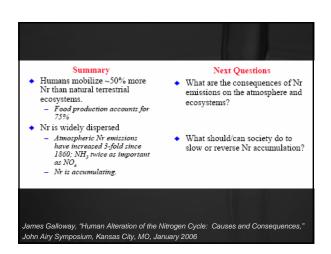
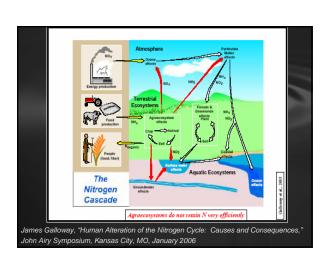
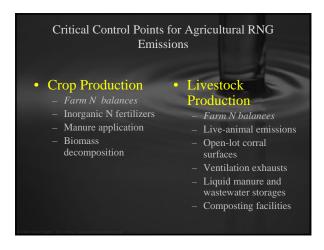


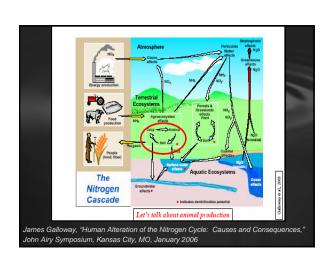
### My Assignment

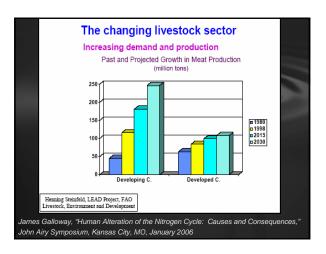
- State of the art for agricultural N emissions
- N emissions control practices
- Costs of control practices and technologies
- *All of that in 30 minutes*



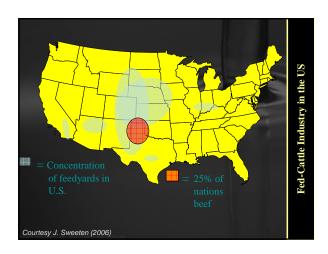




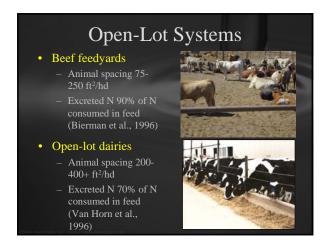








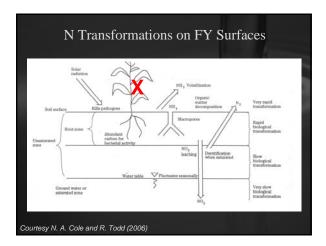




# Fate of Excreted N in Open-Lot Systems Collected in solid manure Spread Stored (stockpiles, mounds, other) Composted and spread Remains on corral surface Stable if it remains dry Runs off into holding pond Volatilized as NH<sub>3</sub>(g) directly Increases with wet/dry cycling

#### NH<sub>3</sub> Loss: Open Lots vs. Ponds

- Open lots
  - Large area source, 2-9 acres per 1,000 head capacity
  - Variable emissions driven by wet/dry cycles, short-term temperature fluctuations
- Lagoons and holding ponds
  - Much smaller area source, 1-10 acres total
  - Seasonal temperature fluctuations
  - Continuous releases; f(temp, wind speed, RH)



### NH<sub>3</sub> Concentrations Near Alberta Feedyards

- Alberta Environment (2000)
  - One-hour average concentrations
  - Up to  $\sim 800 \, \mu \text{g m}^{-3} \, \overline{\text{NH}_3 \text{N}}$
- McGinn *et al.* (2003)
  - Daily averages of 5-minute concentrations
  - Two highest values on days of lowest wind speeds
  - Up to  $\sim 1,500 \, \mu g \, m^{-3} \, NH_3 N$

#### • NH<sub>3</sub> presents steep challenges because of its:

- High reactivity with anions and surfaces
- High aqueous solubility
  - Deposition
  - Condensation
- Kinetically limited redox pathways w/NO<sub>x</sub> species
- Numerous pools and pathways in real systems
- Sensitivity to pH
- Accounting for all of those factors in a single measurement scheme is complicated
- Uncertainty analysis assumes all sources of bias (systematic error) have been eliminated

### NH<sub>3</sub> Flux Estimates by 5 Methods

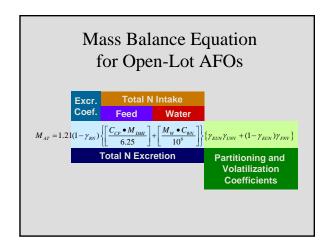
Courtesy N. A. Cole and R. Todd (2006)

### The Holy Grail

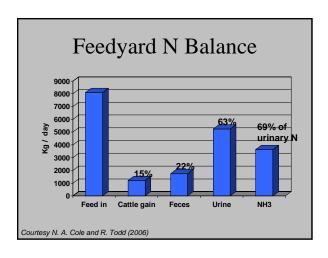
A range of emission factors that expresses the most probable, scientifically justifiable, annualized, NH<sub>3</sub> emission flux from feedyards and dairies as a function of herd size, stocking density or other appropriate measure of capacity or throughput

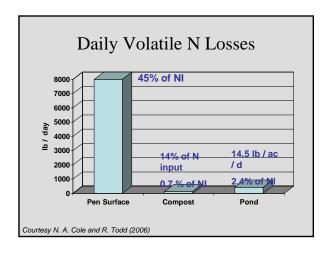
### Where We Are Today There are dozens of different ways of estimating NH<sub>3</sub> flux from an open-lot AFO Today, we consider results from several of them Getting at the *true flux* requires a convergence of results from independent





methods, but even that's not enough





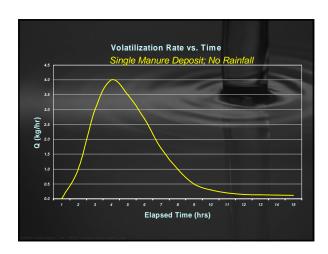
Method	Beef	Dairy	Comments
	% of Fed N		
N Balance	44	<80	Uncertainty analysis nearly complete (beef)
N:P Ratio	48		Varies from 20-51% depending on source material (fresh manure, pen surface, compost)
Flux Chamber	18	3 (OL) - 5 (FS)	Herds are ~15% dry cows, ~85% lactating; excreted N is 79% of fed N
Flux- Gradient	43		Uncertainty analysis underway
bLS/OPL	41		Uses open-path lasers to measure N
Box Model	31-55		

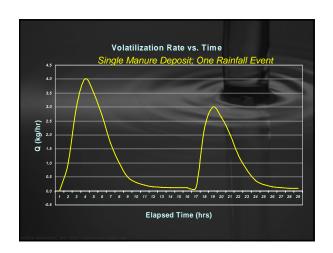


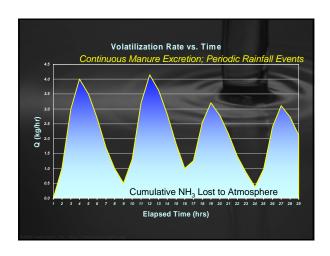
## Urine-Spot NH<sub>3</sub> Emissions The vast majority of NH<sub>3</sub> emissions comes from urine spots Surface chemistry changes rapidly Accurate measurements of NH<sub>3</sub> (and CH<sub>4</sub>, NO<sub>x</sub>) flux are needed to develop appropriate models and make valid treatment comparisons

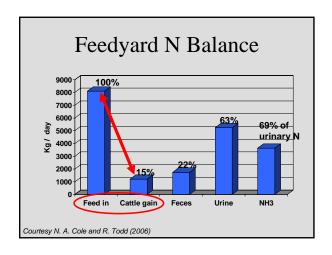






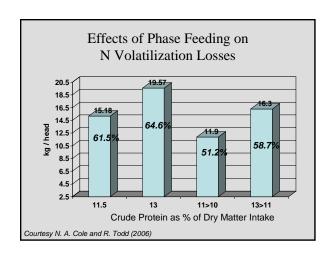






### Reducing Open-Lot NH<sub>3</sub> Flux

- Wet/dry cycles:  $Stop\ H_2O$  applications, improve corral drainage
- Low C:N ratio of manure: Add carbonaceous bedding, mulch or liquid source of organic C (e. g., humates)
- Favorable pH (>7.0): *Alum or other*
- Enzyme-mediated hydrolysis of urea: Urease
- NH<sub>4</sub><sup>+</sup> highly soluble, mobile: *Add strong* adsorption sites (e. g., clinoptilite)



### Surface Amendments

- Shi et al. (2001) *in vitro* evaluations of simulated feedyard surfaces

  - NBPT suppression of urea hydrolysis to NH<sub>4</sub><sup>+</sup> CaCl<sub>2</sub> cation exchange
- Measured cumulative loss over 21 days
- Incremental benefit computed as equivalent N fertilizer maintained in manure; rises and falls with NG/anhydrous prices
- Does not factor in the presence of carbonaceous bedding as is common in Alberta feedyards

### Results of Shi et al. (2001)

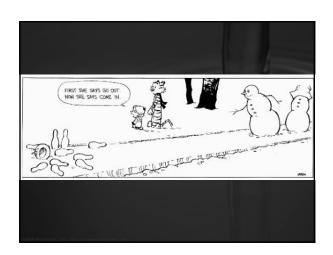
- Alum: 92% reduction at 4,500 kg/ha; B/C=0.17
- CaCl2: 71% reduction at 4,500 kg/ha; B/C=0.16
- NBPT: 65% reduction at 1 kg/ha; B/C=1.75
- Humates: 65% reduction at 9,000 kg/ha; B/C=0.04

### **Surface Amendments**

- Replicating NBPT success outside the laboratory has been unsuccessful so far
- Keeping N as urea in manure surface would increase N pool and require increasing application rates over time
- Urea in solid manure can reduce NO<sub>x</sub> emissions from coal-fired power plants during reburn

### Reducing Open-Lot NH<sub>3</sub> Losses

- Wet/dry cycles: Stop H<sub>2</sub>O applications, improve corral drainage
- Low C:N ratio of manure: Add carbonaceous bedding, liquid C source (humates; dilute beet extract?) or mulch
- Favorable pH (>7.0): *Alum or other*
- Enzyme-mediated hydrolysis of urea: *Urease inhibitors*
- NH<sub>4</sub><sup>+</sup> highly soluble, mobile: *Add strong adsorption sites* (e. g., clinoptilite)
- Extensive area source: Manure harvesting







### A Reminder to Alberta's Policymakers We can design innovative stuff... ... but can we afford it at current levels of energy use? What about at future levels?